



Isolation, identification and characterization of microbes capable for bio remedial activities isolated from Jharia coalfield, Jharkhand

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Abstract

The present study focused on identifying heavy metal degrading bacteria in soil sample collected from jharia coalfield sites Jharkhand. For isolation of bacteria from the soil sample was serially diluted tenfold; the highest dilution (10^{-6}) sample was inoculated onto BBH agar plates and incubated at 37°C for 24 hrs. Colonies with diverse morphological features from the mixed cultures were sub cultured in order to obtain pure colonies. These colonies were then grown on nutrient agar containing diverse molarities of salts of heavy metals (Pb, Cu, Zn and Mn). The highest concentration of each heavy metal growth was observed were as follows: Zn-8mM, Cu-4mM, Pb-6mM and Mn-4mM. The bacteria having multiple tolerances were identified as *Bacillus spp.* on the basis of colony morphology, microscopic examination and biochemical tests. *Bacillus licheniformis* was identified through FAME GC. The tolerant isolate was further subjected to *in vitro* antibiotic cultural sensitivity assay and were found resistant to multiple antibiotics. The study is of significance in that such bacteria can be utilized for bioremediation of heavy metals in the environment particularly in the treatment of heavy metal contaminated soil.

Keywords: *Bacillus licheniformis*, bioremediation, antibiotic, heavy metal

Introduction

Industrial wastewater is a chief source of heavy metal pollution in environment. Heavy metals are economic importance in industrial utilize as well as the largest part of pollutants in the environment. Heavy metals have become a serious hazard to living organisms in environment (C. Su, 2014) [3]. Metal toxicity is large environmental distress because of their nonbiodegradability and bioaccumulation in nature (R. K. Gautam *et al.*, 2014) [7]. A number of inorganic metals such as nickel (Ni), magnesium (Mg), chromium (Cr³⁺), copper (Cu), manganese (Mn), calcium (Ca), sodium (Na) and zinc (Zn) are essential elements required in tiny number for redox and metabolic functions. Heavy metals like lead (Pb), aluminium (Al), cadmium (Cd), mercury (Hg), gold(Au) as well as silver (Ag) do not have any toxic and biological role to living organisms (R. Turpeinen *et al.*, 2002) [8].

Bioremediation is engaged in categorize to convert toxic heavy metals into a less harmful condition via microbes or its enzymes to clear out polluted environment (A. Akcil *et al.*, 2015) [1]. Bioremediation is cost effective and environmentally friendly in the renewal of the environment (Y.Ma,M *et al.*, 2016) [12]. Bioremediation of heavy metals has some limits. These are production of lethal metabolites by nonbiodegradability and microbes of heavy metals. The straight use of microorganisms with characteristic of catabolic potential and their products like bio surfactant and enzymes is a new approach to increase and enhance their remediation efficacy (T. T. Le *et al.*, 2017) [9]. Dissimilar alternatives have also been probable to enlarge the

microbiological techniques towards the remediation of heavy metals. Bio film mediated bioremediation could be useful for cleaning up of heavy metal infected environment.

Microbial-metal communications is mainly paying attention on metals elimination such as depollution and remediation. The current stimulation of the utilize solid-state electrodes as electron acceptors and donors for microbial growth has brought modern scenario, resulting to microbial-electrochemical technologies (METs) (J. C. Thrash and J. D. Coates, 2008) [4].

Microorganisms used as a green approach for the production of metallic nanoparticles (NPs) have been reported (T. Klaus-Joerger *et al.*, 2001) [10]. Hereditarily modified microorganisms have also been using for remediation technique (V. Paliwal *et al.*, 2012) [11]. Chemical modification and Genetic engineering could modify the mechanism of cells surface and could professionally progress the adsorption ability as well as selectivity to target-metal species. Numerous factors which limit and influences bioremediation efficiency consist of redox, temperature, nutritional status, pH, chemical composition and potential moisture of heavy metals (P. K. Shukla, *et al.*, 2013) [5]. Microbes alone have exposed imperfect efficiency remaining to different factors together with excessive and poor competitiveness heavy metal concentrations. Effectiveness could be improved by numerous amendments with biosurfactants, inorganic nutrients, bulking agents, compost and biochar (A. Wiszniewska *et al.*, 2016) [2].

Therefore, the main goal of this study was to get a strain Collection of heavy metal degrading bacteria able to apply wonderful degradation possible against high levels of heavy metal pollutants. Furthermore, our aim was to determine the heavy metal tolerance and antibiotic resistance of isolated and identified heavy metal-degrading bacterial strains.

Materials and Methodology

Soil sampling

Soil samples were collected from different areas of Jharia coalfield Jharkhand. Samples were collected during the month of June. The top 15 cm of soil was collected into sterile plastic bags for microbiological analysis. Samples were stored at 4°C until further processing.

Purification and culturing of heavy metal degrading bacteria

Heavy metal degrading bacteria were isolated using enrichment containing: 40 ml BBH mineral broth medium supplemented with 0.1mM heavy metals and 4.0 g of contaminated soil sediment. And incubation at 32°C for 2 weeks, the enriched cultures were serially diluted and inoculated onto BBH agar plates. Colonies with diverse morphologies were preferred as candidate heavy metal degrading strains and were maintained on standard Nutrient Agar (Hi Media).

Qualitative analysis (Dye method)

10 isolates were checked for the degradation of heavy metals. The setup was prepared in 250 ml sterilized conical flasks. Solid heavy metals (10 g/l) were dissolved in dichloromethane and filtered sterilized through 0.22 µm membrane. 2 ml of solution at a final concentration of 100 ppm were added in different conical flasks and left open in sterilized conditions for the solvent gets evaporated. Than 100 ml of BBH media with trace elements and 100 µl of resazurin was added in the flasks. 500 µl of bacterial culture was added (OD₆₀₀=0.5) in BBH broth in the flasks and kept in shaking incubator at 120 rpm and 30°C. Solvent, media and resazurin was used as control 1 and flask with inoculant and media was used as control 2.

Resazurin, which is also preferred to as Alamar blue. Alamar blue is a non-toxic dye condensed intracellular to resofurin through enzymes in the electron transport system. In the beginning a blue coloured resazurin is change into a pink colored resofurin through the loss one atom of oxygen. In second stage of reduction the pink colored resofurin is further more reduced to colorless hydrosesofurin

Calculations

Formula used to calculate the residual concentration and % degradation are given below:

Residual concentration (in ppm)

$$= \frac{\text{Sample Area}}{\text{Standard Area}} \times \frac{\text{Amount of std. injected (ng)}}{\text{Vol. of sample taken (ml)}} \times \frac{\text{Sample make up volume}}{\text{Volume of sample injected}} \times \text{DF}$$

$$\% \text{Degradation} = \frac{\text{Initial Concentration} - \text{Residual concentration}}{\text{Initial Concentration}} \times 100$$

Fame GC Analysis

The Sherlock TM Chromatographic Analysis System (CAS) automatically identifies fatty acid methyl esters (FAMES) by gas chromatography. The FAME profile can be used for fatty acid compound identification (marine oil analysis, PLFA analysis), or the entire fatty acid profile can be compared by pattern recognition to one of Sherlock's microbial libraries (e.g. environmental aerobic bacteria).

Libraries & Methods

The primary Sherlock methods and libraries available include:

Fatty Acid Compound Naming and Quantitation

- Microbial Community Analysis (PLFA) - 150 soil phospholipid fatty acids
- Marine Oil Analysis (MARINE) - 147 marine fatty acids, including Omega-3s, Omega-6s and Omega-9s

Sample Preparation

With inexpensive reagents, available from almost any chemical supply house, a technician averages 5 minutes per sample to prepare a batch of 30 samples. Each sample is prepared for analysis using a liquid-liquid extraction in a single test tube.

1. Harvesting a small quantity (~20mg cells) from the culture plate is the most labor-intensive step. It will typically take 1 hour or less to harvest cells from 30 plates into 30 test tubes.
2. The four-step liquid-liquid extraction process requires about 1½ hours or less for a batch of 30 samples. During the extraction process, there are approximately 35 minutes of "wait time" available for the technician to do paper work and other tasks.
3. The sample preparation is the same, regardless of microbial type. It is not necessary to do a Gram stain or other offline tests before preparing and analyzing a sample.

Gases

The Sherlock requires a specific type and quality of gases to function properly:

Carrier Gas

1. Hydrogen, 99.999%+, 150 cc/min+
2. Note: Helium cannot be used

Makeup Gas

1. Nitrogen, 99.999%+
2. Industrial Grade Air, < 1ppm THC

Testing heavy metal and antibiotic resistance of strains

Heavy metal resistance of strains was tested in nutrient broth tubes containing different concentrations (0.25, 0.5, 1.0, 2.0, 1.0, 2.0, 3.0, 4.0, 5.0, 6.0, 7.0, 8.0 mm) of CuSO₄.5H₂O, Pb (NO₃)₂, ZnSO₄.7H₂O, MnSO₄.H₂O, C₈H₄K₂O₁₂Sb₂.3H₂O, SnCl₂.2H₂O (Hi Media). Stock solutions were prepared in deionized water and were also filter sterilized before inoculation. All tests were performed in triplicates and the results were evaluated visually for growth against heavy-metal free control cultures and blank tubes.

Antibiotics sensitivity of selected strains was performed by the Kirby Bauer's disc diffusion method using Muellere Hinton agar (Bauer *et al.*, 1996; CLSI, 2013). Cephalosporins (cefuroxime

sodium-CXM30, cefotaxime-CTX30, cefoperazone-CFP75), penicillins (piperacillin-PRL100, penicillin G-10U, amoxyclav-30), tetracyclines (tigecycline-TGC15), carbapenems (imipenem-IPM10), quinolones (norfloxacin-NOR10). All the experiments were performed in triplicates.

Result

Purification and culturing of heavy metal degrading bacteria

5 diverse appearing colonies were recognized which were maintained in Agar slants after that enriching them from heavy metal contaminated soil samples.



Fig 1: Bacterial cultures isolated from soil sample.

Table 1: Morphological Characteristics of Isolates

| Characteristics | JCJ-1 | JCJ-2 | JCJ-3 | JCJ-4 | JCJ-5 |
|--------------------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| Colonial characteristics | Circular, white | Circular, white | Circular, white | Circular, white | Circular, white |
| Gram's reaction | + | + | - | - | + |
| Shape | Circular | Rod | Rod | Circular | Filamentous |
| Catalase | + | + | + | - | + |
| Oxidase | + | + | + | + | + |
| Methyl Red | - | - | - | - | - |
| Voges roskauser | + | + | + | - | - |
| Indole | - | - | - | - | - |
| Citrate | - | + | + | + | + |

Qualitative degradation of the isolates (Dye method)

The heavy metal degrading of the 5 isolates by dye method is presented below tables. *Bacillus spp.* showed greatest microbial activity by dropping resazurin to pink colored resorufin. This one isolate was further applied for molecular identification, quantitative heavy metal degradation, antibiotic as well as heavy metal resistance.



Fig 2: Resazurin (purple) and resorufin (pink) formation

Table 2: Qualitative degradation against Pb

| SN. | Bacterial species | Sorption Efficiency (%) |
|-----|-------------------|-------------------------|
| 1 | Micrococcus | 10.11 |
| 2 | Bacillus | 78.32 |
| 3 | Pseudomonas | 78.9 |
| 4 | Staphylococcus | 18.9 |
| 5 | streptomyces | 29.5 |

Table 3: Qualitative degradation against Cu

| SN. | Bacterial species | Sorption Efficiency (%) |
|-----|-------------------|-------------------------|
| 1 | Micrococcus | 35.78 |
| 2 | Bacillus | 67.33 |
| 3 | Pseudomonas | 67.33 |
| 4 | Staphylococcus | 76.8 |
| 5 | streptomyces | 29.5 |

Table 4: Qualitative degradation against Zn

| SN. | Bacterial species | Sorption Efficiency (%) |
|-----|-------------------|-------------------------|
| 1 | Micrococcus | 2.11 |
| 2 | Bacillus | 59.8 |
| 3 | Pseudomonas | 46.9 |
| 4 | Staphylococcus | 1.11 |
| 5 | Streptomyces | 11.12 |

Table 5: Qualitative degradation against Cr

| SN. | Bacterial species | Sorption Efficiency (%) |
|-----|-------------------|-------------------------|
| 1 | Micrococcus | 38.78 |
| 2 | Bacillus | 49.2 |
| 3 | Pseudomonas | 98.5 |
| 4 | Staphylococcus | 46.2 |
| 5 | Streptomyces | 73.1 |

Identification of the bacterial strain

The unknown bacterial culture was identified as *Bacillus-licheniformis* through FAME GC (Table 6).

Table 6: *Bacillus-licheniformis*

| RT | Response | Ar/Ht | RFact | ECL | Peak Name | Percent | Comment1 | Comment2 |
|--------|----------|-------|-------|---------|--------------|---------|---------------------|------------------|
| 0.7599 | 1.59E+6 | 0.006 | ---- | 6.6052 | | ---- | < min rt | |
| 0.7733 | 1.411E+9 | 0.021 | ---- | 6.6907 | SOLVENT PEAK | ---- | < min rt | |
| 1.0041 | 856 | 0.009 | ---- | 8.1775 | | ---- | < min rt | |
| 1.1015 | 842 | 0.015 | ---- | 8.7922 | | ---- | < min rt | |
| 1.2309 | 369 | 0.008 | 1.173 | 9.6099 | 10:0 iso | 0.10 | ECL deviates -0.006 | Reference -0.014 |
| 1.4284 | 2065 | 0.010 | ---- | 10.6805 | | ---- | | |
| 1.5016 | 738 | 0.012 | ---- | 11.0392 | | ---- | | |
| 1.7369 | 7435 | 0.009 | 1.045 | 12.0139 | 12:0 | 1.84 | ECL deviates 0.014 | Reference 0.017 |
| 1.9185 | 4505 | 0.011 | ---- | 12.6744 | | ---- | | |
| 2.1970 | 4450 | 0.009 | 0.989 | 13.6286 | 14:0 iso | 1.04 | ECL deviates 0.001 | Reference 0.006 |
| 2.3084 | 2537 | 0.009 | 0.979 | 13.9996 | 14:0 | 0.59 | ECL deviates 0.000 | Reference 0.006 |

| | | | | | | | | |
|--------|--------|-------|-------|---------|------------------|-------|----------------------|----------------------|
| 2.5078 | 115394 | 0.009 | 0.964 | 14.6307 | 15:0 iso | 26.32 | ECL deviates -0.001 | Reference 0.005 |
| 2.5372 | 161683 | 0.009 | 0.962 | 14.7238 | 15:0 anteiso | 36.81 | ECL deviates -0.001 | Reference 0.005 |
| 2.6230 | 1194 | 0.012 | ---- | 14.9952 | 15:0 | ---- | ECL deviates -0.005 | |
| 2.7576 | 3433 | 0.010 | 0.950 | 15.4111 | 16:1 w7c alcohol | 0.77 | ECL deviates -0.003 | |
| 2.8282 | 19649 | 0.009 | 0.946 | 15.6292 | 16:0 iso | 4.40 | ECL deviates -0.004 | Reference 0.001 |
| 2.8765 | 4104 | 0.011 | 0.944 | 15.7785 | 16:1 w11c | 0.92 | ECL deviates -0.003 | |
| 2.9465 | 14037 | 0.009 | 0.941 | 15.9950 | 16:0 | 3.13 | ECL deviates -0.005 | Reference -0.001 |
| 3.0285 | 565 | 0.012 | 0.938 | 16.2490 | 15:0 2OH | 0.13 | ECL deviates -0.006 | |
| 3.0647 | 627 | 0.010 | ---- | 16.3612 | | ---- | | |
| 3.0815 | 4097 | 0.009 | 0.936 | 16.4133 | 17:1 iso w10c | 0.91 | ECL deviates -0.001 | |
| 3.1124 | 5188 | 0.010 | 0.935 | 16.5090 | Sum In Feature 4 | 1.15 | ECL deviates -0.003 | 17:1 anteiso B/iso I |
| 3.1264 | 2870 | 0.009 | 0.935 | 16.5524 | 17:1 anteiso w9c | 0.64 | ECL deviates 0.000 | |
| 3.1532 | 30805 | 0.009 | 0.934 | 16.6354 | 17:0 iso | 6.81 | ECL deviates -0.002 | Reference 0.001 |
| 3.1846 | 58557 | 0.009 | 0.933 | 16.7327 | 17:0 anteiso | 12.93 | ECL deviates 0.000 | Reference 0.002 |
| 3.4742 | 411 | 0.009 | 0.927 | 17.6350 | 18:0 iso | 0.09 | ECL deviates -0.001 | Reference -0.001 |
| 3.5258 | 2702 | 0.014 | 0.926 | 17.7960 | 18:1 w9c | 0.59 | ECL deviates 0.002 | |
| 3.5920 | 1324 | 0.011 | 0.926 | 18.0026 | 18:0 | 0.29 | ECL deviates 0.003 | Reference 0.001 |
| 3.6528 | 884 | 0.012 | 0.925 | 18.1969 | 17:0 iso 3OH | 0.19 | ECL deviates 0.004 | |
| 3.6876 | 1915 | 0.014 | ---- | 18.3082 | | ---- | | |
| 3.7695 | 597 | 0.013 | 0.924 | 18.5702 | 17:0 3OH | 0.13 | ECL deviates 0.005 | |
| 3.7996 | 2232 | 0.012 | ---- | 18.6663 | | ---- | | |
| 3.8231 | 1049 | 0.015 | 0.924 | 18.7415 | 19:0 anteiso | 0.23 | ECL deviates 0.004 | Reference -0.002 |
| 4.0129 | 985 | 0.024 | ---- | 19.3574 | | ---- | | |
| ---- | 5188 | --- | ---- | ---- | Summed Feature 4 | 1.15 | 17:1 iso I/anteiso B | 17:1 anteiso B/iso I |

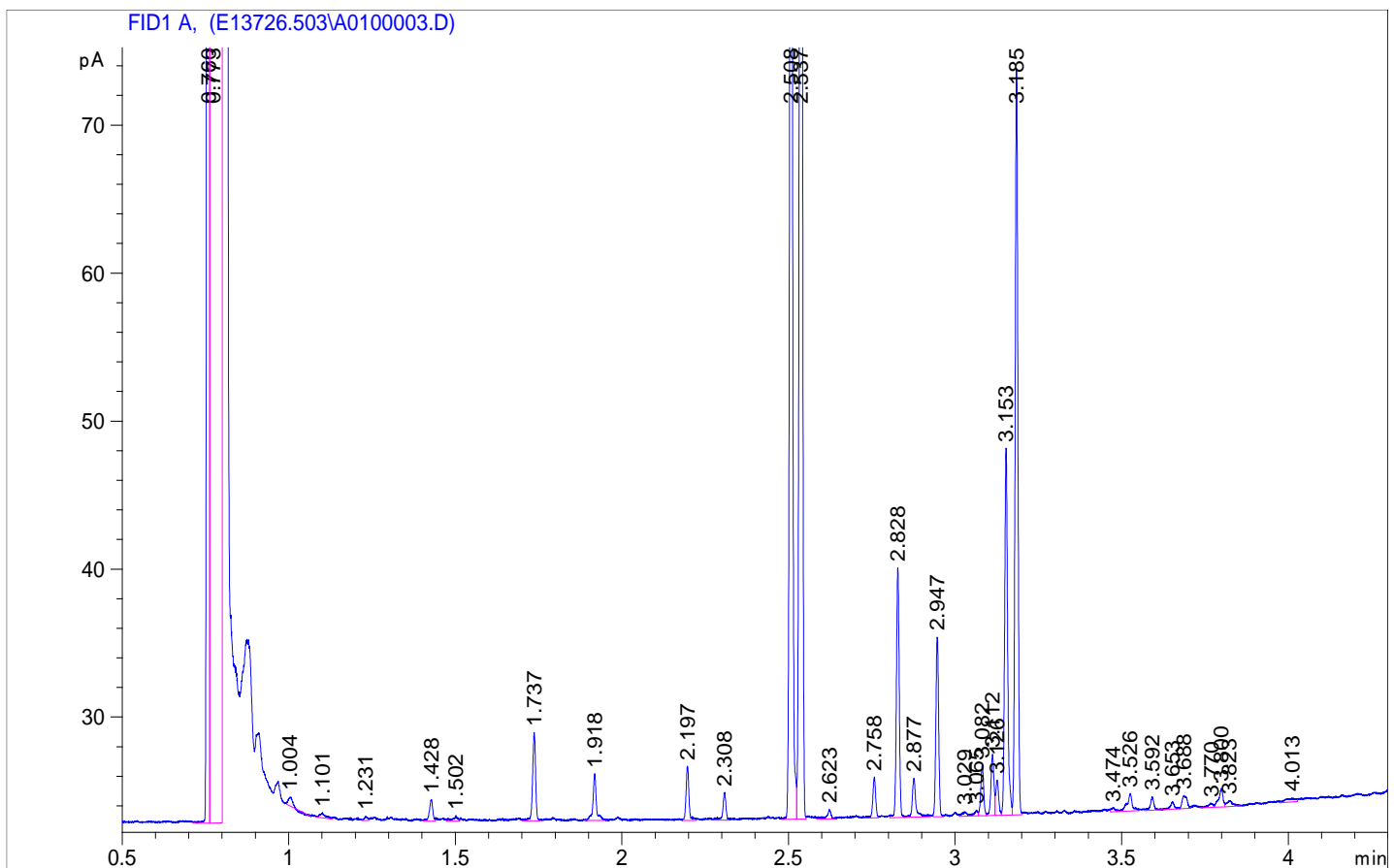


Fig 3: Chromatogram of *Bacillus-licheniformis*

Heavy metal resistance of identified bacteria

Maximum tolerance was shown by *Bacillus licheniformis* against (15 mM) and least tolerance was shown by four bacterial species

except *B. licheniformis* MTC values of 0 mM for Stannous Chloride. *Staphylococcus spp.* showed least tolerance against Cu⁺ (0.25 mM) but was able to tolerate 10 mM of Mn²⁺.

Table7: Maximum tolerated concentration (MTC) values of tested heavy metals

| Strains | MTC values of heavy metals (mm) | | | | | |
|----------------|---------------------------------|------------------------------|--------------------------------|----------------------------|-------------------------------|--|
| | Lead Nitrate (PbNO3) | Copper Sulphate (CuSO4.5H2O) | Stannous Chloride (SnCl2.2H2O) | Zinc Sulphate (ZnSO4.7H2O) | Manganese Sulphate (MnSO4H2O) | Potassium Antimony Tartarate (C8H4K2O12Sb2.3H2O) |
| Micrococcus | 5 | 3 | 3 | 5 | 9 | 6 |
| Bacillus | 7 | 4 | 4 | 8 | 11 | 16 |
| Pseudomonas | 6 | 4 | 4 | 5 | 3 | 5 |
| Staphylococcus | 6 | 4 | 4 | 6 | 8 | 6 |
| Streptomyces | 5 | 5 | 4 | 5 | 6 | 7 |

Antibiotic resistance of strains

All identified bacteria showed resistance against penicillin G and cefuroxime, impenem, amoxyclav. *B. licheniformis* showed

resistance against 4 out of 6 antibiotics and the other four (Micrococcus, Pseudomonas, Staphylococcus, Streptomyces) were resistant only to 2 out of 6 antibiotics used (Table 8).

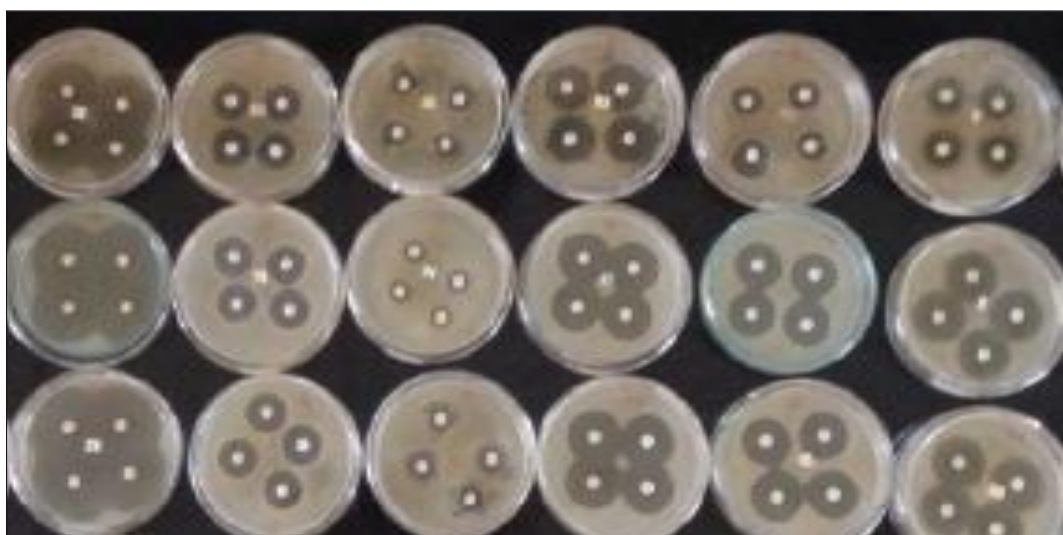


Fig 4: Antibiotic sensitivity pattern of the *Bacillus licheniformis*.

Table 8: Susceptibility of identified strains against antibiotics

| Antibiotic & concentration → | Penicillin G | Piperacillin | Norfloracin | Imipenem | Cefotaxime | Amoxyclav |
|------------------------------|--------------|--------------|-------------|-------------|-------------|-------------|
| Strains ↓ | | | | | | |
| Micrococcus | 9.3 ± 0.57 | 10.3 ± 0.56 | 7.3 ± 0.55 | 13 ± 0 | 09.6 ± 0.58 | 15.3 ± 0.58 |
| Bacillus | 25 ± 0.1 | 25.4 ± 0.5 | 28.6 ± 0.58 | 40.3 ± 0.58 | 26 ± 1 | 30.6 ± 0.58 |
| Pseudomonas | 13.6 ± 0.58 | 14.3 ± 0.58 | 16.6 ± 0.58 | 40.3 ± 1.15 | 22.3 ± 0.58 | 29 ± 1 |
| Staphylococcus | 2.6 ± 0.58 | 7.6 ± 0.58 | 2.6 ± 0.58 | 4.6 ± 0.58 | 15 ± 0 | 15 ± 0 |
| Streptomyces | 2.3 ± 1.1 | 2 ± 0 | 3.3 ± 0.58 | 4 ± 0 | 2 ± 1 | 3 ± 0 |

Conclusion

The present study was of the bacterial bioremediation characters and their heavy metal and antibiotic tolerance from the heavy metal contaminated soil located in Jharia Coalfield Jharkhand, India. Mainly strong strains characterized were predominantly from the *Bacillus licheniformis* which showed mixed resistance against antibiotics and heavy metals. Highest degradation was observed by *B. licheniformis*. *Bacillus licheniformis* strain showed resistance against most of heavy metals and the most active bacteria to degrade the selected heavy metals. All the isolates except *Bacillus licheniformis* were resistance against 4 antibiotics out of 6 antibiotics and other bacterial spp. showed less sensitive to antibiotic resistance.

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